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Reactivity of 4-Bromomethylquinoline Derivatives Towards Glycine and Thioglycolic Acid. A New Ring System

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Condensation of 4-bromomethylquinoline derivatives 1a-1c with glycine and thioglycolic acid gave the corresponding quinolylmethylglycine and quinolylmethylthioacetic acid derivatives 2a-2c and 2d-2f, respectively. Cyclization of 2a-2f was affected either by polyphosphoric acid or concentrated sulphuric acid to give 3a-3f. Chlorination of 2a-2f and 3a-3f were also accomplished.

(Keywords: Benzo[c]-2,6-naphthyridine; Bromomethylquinoline; Nitrogen heterocycles; Thiopyrano[4,3-c]quinoline)

Die Reaktivität von 4-Brommethylchinolin-Derivaten gegenüber Glycin und Thioglycolsäure. Ein neues Ringsystem

Die Reaktion von 4-Brommethylchinolin-Derivaten 1 a - c mit Glycin und Thioglycolsäure gab die entsprechenden Kondensationsprodukte 2a - c und 2d - f. 2a - f konnten mit Polyphosphorsäure bzw. mit konzentrierter Schwefelsäure zu Benzo[c]-2,6-naphthyridinen 3a - f zyklisiert werden. 2a - fund 3a - f waren einer Chlorierung (POCl₃) zugängig.

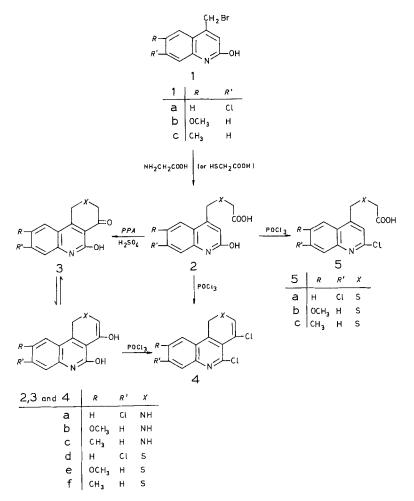
Certain substituted benzonaphthyridine derivatives are known to possess antimalarial activity¹. This prompted us to endeavour the synthesis of some benzo[c]-2,6-naphthyridines (known till 1970 as benzo[b]-2,6-naphthyridine) and its thia-analogue, 1H-thiopyrano[4,3-c]-quinolines, since there is only limited knowledge concerning these isomers.

The first observation concerning the isomer, benzo[c]-2,6-naphthyridine was made during degradation of the alkaloid "calycanthine" which affords dibenzo[c,f]-2,6-naphthyridine (calycanine)²⁻⁴.

The parent 2,6-naphthyridine was first synthesized by a long reaction sequence^{5,6} whereas in the present work the naphthyridine derivatives 3a-3f and 4a-4f were synthesized by only two steps.

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Thus the 4-bromomethylquinoline derivatives⁷ 1 a-1 c reacted with glycine or thioglycolic acid in alkaline medium to give the corresponding quinolylmethylglycine derivatives 2 a-2 c and quinolylmethyl-thioacetic acid derivatives 2 d-2 f, respectively, whereas an attempted



condensation of **1 a-1 c** with glycolic acid under different conditions was unsuccessful. The IR spectrum of **2e** shows signals at 1,715 (C=O carboxylic), 1,610 (C=N) and at 750 cm⁻¹ (--C-S-). The characteristic feature of the IR spectrum of **2b** is the presence of a signal at 3,340 cm⁻¹ (NH stretching).

Cyclization of 2a-2f either by concentrated sulphuric acid or polyphosphoric acid afforded the desired dihydro[c]-2,6-naphthyridin-

4(1H)-one derivatives $3a-3c^8$ and the new ring system, 1H-thiopyrano[4,3-c]quinolin-4(3H)-one derivatives 3d-3f, respectively. These derivatives 3a-3f are present almost entirely in the enol form as revealed by spectroscopic data.

Thus the IR spectrum for **3a** shows absorption bands at 3,250 (NH), 1,660 (C=O), 1,610, 1,550 cm⁻¹ (C=N, C=C aromatic) and at 790 cm⁻¹ (C—Cl aromatic). The NMR spectrum of **3a** (CF₃COOD) shows the presence of absorption signals at $\delta 2.15$ (2 H, CH₂ group protons in position-1, singlet), $\delta 5.25$ (1 H, OH proton, singlet) and at $\delta 7.40$ -8.25 ppm (4 H, aromatic protons, multiplet), as main characteristic features. Also, the NMR spectrum of **3d** (CF₃COOD) shows signals at $\delta 2.18$ (1 H, CH proton of the heterocyclic sulphur ring, singlet), $\delta 2.55$ (3 H, CH₃ group protons, singlet), $\delta 4.25$ (2 H, CH₂ group protons of the heterocyclic sulphur ring, singlet), $\delta 7.25$ (1 H, OH proton, singlet) and a multiplet centered at $\delta 7.90$ ppm (3 H, aromatic protons).

Treatment of either 2a-2c or 3a-3c with phosphorous oxychloride gave the corresponding 4,5-dichloro-2-hydro(1*H*)-benzo[*c*]-2,6-naphthyridine derivatives 4a-4c, respectively. But if 2d-2f was treated with phosphorous oxychloride, a mixture of the corresponding 2chloroquinolylmethylthioacetic acid derivatives 5a-5c and 4,5-dichloro-1*H*-thiopyrano[4,3—*c*]quinoline derivatives 4d-4f was obtained. Separation of the mixture was achieved by treatment with alkali. Furthermore, 4d-4f were obtained through chlorination of 3d-3f with phosphorous oxychloride. The IR spectrum of 5b shows a peak at 1,705 cm⁻¹(COOH).

Because of the known biological activity of the N,N-diethylaminoethylamino group⁹ we have also prepared the diamino derivatives **6a** and **6b** by condensation of **4a** and **4d** with N,N-diethylaminoethylamine. The IR spectrum of **6b** shows absorption bands at 3,350 (NH), 1,610, 1,550 (C=N, C=C aromatic) and at 785 cm⁻¹ (C-Cl aromatic).

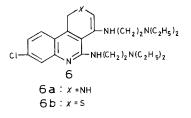
Experimental

All melting points (°C) were uncorrected and were taken in a Gallenkamp electric melting point apparatus and Boetius melting point microscope. IR spectra were performed on a Carl-Zeiss Jena Infracord Spectrophotometer model "UR 10" using KBr. NMR spectra were obtained in deuterotrifluoroacetic acid solution with a Varian model "A-60". Elementary analyses (C, H, N) of 2a-f, 3a-f, 5a-c and 6a, b were in good agreement with the proposed structures.

N[(7-Chloro-2-hydroxy-4-quinolyl)methyl]glycine (2 a)N[(2-Hydroxy-6-methoxy-4-quinolyl)methyl]glycine (2 b) andN[(2-Hydroxy-6-methyl-4-quinolyl)methyl]glycine (2 c)

To a mixture of glycine $(37.5\,\mathrm{g}; 0.5\,\mathrm{mol})$ and 4-bromomethylquinoline derivatives $1\,\mathrm{a-1c}$ (0.1 mol) were added 200 ml of a 5N-NaOH solution portionwise over a period of 2h, with stirring, then refluxed for further 10 h

and filtered. The filtrate was neutralized with dilute hydrochloric acid and filtered again. The white precipitate was washed with water and recrystallized from dilute acetic acid to give 2a-2c in 30-33% yield. M. p. 274° (2a), 227° (2b), 218° (2c).



8-Chloro-2,3-dihydro-5-hydroxybenzo[c]-2,6-naphthyridine-4(1H)-one ($\mathbf{3a}$), 2,3-dihydro-5-hydroxy-9-methoxybenzo[c]-2,6-naphthyridine-4-(1H)-one ($\mathbf{3b}$) 2,3-dihydro-5-hydroxy-9-methylbenzo[c]-2,6-naphthyridine-4-(1H)-one ($\mathbf{3c}$)

Method (A)

The glycine derivatives 2a-2c (5g) were dissolved in 25 ml concentrated sulphuric acid heated on a steam bath for 2h, cooled, poured into 250 ml ice water and filtered. The yellow precipitate was washed with dilute ammonia then with water, dried and recrystallized from acetic acid to give 3a-3c in 75-80%yield.

Method (B)

2a-2c (2g) were refluxed in 25g of polyphosphoric acid at 140° for 6h, cooled, poured into 200 ml ice water, neutralized with dilute ammonia and filtered. The precipitate was washed with water, dried and recrystallized from acetic to give 3a-3c in ca. 85% yield. M. p. 223° (3a), 270° (3b), 312° (3c).

2-Hydroxy-4,5,8-trichloro-1H-benzo[c]-2,6-naphthyridine (4 a). 4,5-Dichloro-2-hydro-9-methoxy-1H-benzo[c]-2,6-naphthyridine (4 b) and 4,5-Dichloro-2-hydro-9-methyl-1H-benzo[c]-2,6-naphthyridine (4 c)

2a-2c (1g) were refluxed gently in POCl₃ (10 ml) under dry conditions for 6 h, cooled, poured carefully into 500 ml icecold ammonia solution (5%) and filtered. The precipitate was washed thoroughly with water, dried in a vacuum desiccator over P₂O₅. The yield was 80-85%. Crude melting points were 184-186° (4a), 128° (4b) and 155-157° (4c). These chloro derivatives were soluble in most organic solvents, but easily hydrolyzed to the corresponding hydroxy derivatives 3a-3c during crystallization. Thus, they were identified from their reactions with diethylaminoethylamine. Treatment of 3a-3c with POCl₃, gave identical products (4a-4c) in 70-75% yield.

8-Chloro-4,5-bis{[2-(diethylamino)ethyl]amino}-2-hydro-1Hbenzo[c]-2,6-naphthyridine (6a)

A mixture of 4a (0.3g; 0.001 mol) and 1g of phenol was refluxed for 2 h on a water bath. Diethylaminoethylamine (0.32g; 0.003 mol) was added, refluxing continued for 3 h, then cooled, poured into cold dilute NaOH solution and filtered. The precipitate was washed thoroughly with water, dried and

recrystallized from ethanol. Pale yellow crystals of 6a were obtained in 55% yield. M. p. 238°.

Thioglycolic acid (13.8 g; 0.15 mol) was added to a solution of 4bromomethylquinoline derivatives 1 a-1 c in ethanol (400 ml); a solution of NaOH (100 ml; 12%) was then portionwise added over a period of 4 h while stirring on a steam bath. The reaction mixture was then cooled, diluted with water and filtered. The filtrate was neutralized with dilute hydrochloric acid and filtered again. The yellow precipitate was washed with water, dried and recrystallized from acetic acid to give 2d-2f in 85-90% yield. M. p. 315° (decomp.; 2d), 293° (2e), 304° (2f).

8-Chloro-5-hydroxy-1H-thiopyrano[4,3-c]quinoline-4-(3H)-one (3d), 5-Hydroxy-9-methoxy-1H-thiopyrano[4,3-c]quinoline-4-(3H)-one (3e) and 5-Hydroxy-9-methyl-1H-thiopyrano[4,3-c]quinoline-4-(3H)-one (3f)

Method (A)

2d-2f (5g) were heated in concentrated sulphuric acid (25 ml) on a steam bath for 2h, cooled, poured into ice cold water and filtered. The yellow precipitate was washed with dilute ammonia and water, dried and recrystallized from acetic acid giving 3d-3f in a quantitative yield.

Method (B)

2d-2f(2g) was heated in polyphosphoric acid (25g) at 140° for 6 h, cooled, poured into ice cold water (200 ml), neutralized with dilute ammonia and filtered. The precipitate was washed with water, dried and recrystallized from acetic acid to give 3d-3f in about 90% yield. M. p. 284° (3d), 325° (3e), 298° (3f).

Chlorination of 2 d-2 f

2d-2f (2g) were refluxed in POCl₃ (10 ml) under anhydrous conditions for 2 h, cooled, poured carefully into 250 ml ice water and filtered. The precipitate contains both **5a**-**5c** and **4d**-**4f**. It was treated with dilute sodium bicarbonate solution and filtered again. The precipitate contains **4d**-**4f**. The filtrate was washed with water, dried and recrystallized from benzene [yield 26% for **5a** and **5c**, 35% for **5b**; m. p. 149° (**5a**), 156° (**5b**), 144° (**5c**].

Treatment of 3d-3f with POCl₃ gave identical products (4d-4f).

8-Chloro-4,5-bis{[(2-diethylamino)ethyl]amino}-1H-thiopyrano-[4,3-c]quinoline (6 b)

A mixture of 4d(0.3g; 0.001 mol) and 1g of phenol was refluxed for 2 h on a steam bath. Diethylaminoethylamine (0.32g; 0.003 mol) was added, reflux was continued for 3 h, cooled, poured into cold dilute sodium hydroxide solution, amd filtered. The precipitate was washed thoroughly with water, dried and recrystallized from ethanol. Pale yellow crystals of **6** b were obtained in 45% yield. M. p. 265° .

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